

# Preparation of Methyl 2,3-Anhydro-4,6-*O*-benzylidene-3-deoxy-3-nitro- $\beta$ -D-allopyranoside and Some $C^2$ -Branched-Chain Derivatives of Methyl 4,6-*O*-Benzylidene-2,3-dideoxy-3-nitro- $\beta$ -D-glucopyranoside

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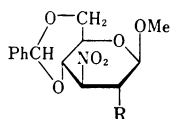
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(Received May 29, 1970)

In a previous paper<sup>1)</sup> we reported that new nucleosides (**3** and **4**) were easily formed on treatment of 3-nitroglucoside **1** with theophylline or 2,6-dichloropurine in the presence of sodium bicarbonate via a two-step mechanism, *i.e.* (i) elimination of acetic acid from **1** to yield  $\alpha$ -nitroolefin **2** and (ii) nucleophilic addition of the purine bases to **2**. Recently Newman and Angier<sup>2)</sup> described  $\alpha$ -nitroepoxides, a new class of compounds, generated from  $\alpha$ -nitroolefins with alkaline hydrogen peroxide.

We have found that on treatment with sodium hypochlorite in THF **2** also gives nitroepoxide **5** as the first member in the field of sugar chemistry. When the solvent was changed from THF to acetone, an unexpected addition of acetone, a very weak nucleophile ( $K_a^{30} 10^{-20}$ ), to **2** predominantly occurred<sup>4)</sup> and a branched-chain compound **6** was obtained.

In THF (20 ml) **2**<sup>5)</sup> (293 mg) was stirred with



**1** R = OAc

**3** R = 7-theophyllinyl

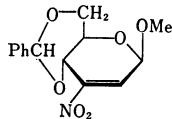
**4** R = 2,6-dichloro-9-purinyl

**6** R = CH<sub>2</sub>-C-CH<sub>3</sub>

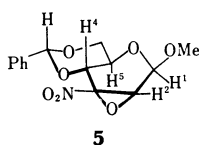
**7** R = CH(CN)<sub>2</sub>

**8** R = CH(CO<sub>2</sub>Et)<sub>2</sub>

**9** R =



**2**



**5**

sodium hypochlorite (*ca.* 20% aq., 1.8 ml) for 2 hr at room temperature. It was then diluted with water (40 ml), condensed *in vacuo* to a half volume, extracted with chloroform, and on evaporation the remaining material was recrystallized from ethanol/water to give in 74% yield epoxide **5**, C<sub>14</sub>H<sub>15</sub>NO<sub>7</sub>,<sup>6)</sup> mp 184–185°C,  $[\alpha]_D^{20} -96.2^\circ$  (*c* 1, MeOH).

When stirred in acetone instead of THF overnight at room temperature, **2** gave in 91% yield adduct **6**, C<sub>17</sub>H<sub>21</sub>NO<sub>7</sub>,<sup>6)</sup> mp 176–177°C (from EtOH),  $[\alpha]_D^{20} -66.9^\circ$  (*c* 1, MeOH), IR(KBr) 1715 cm<sup>-1</sup> (C=O), NMR(CDCl<sub>3</sub>) 7.89 $\tau$  (3H-*s*, Ac).

Similarly, addition of malononitrile ( $K_a^{30} 6.5 \times 10^{-12}$ ) and ethyl malonate ( $K_a^{30} 5 \times 10^{-14}$ ) to **2** afforded the following, in an over 88% yield, respectively: 2-*C*-Dicyanomethyl derivative **7**, C<sub>17</sub>H<sub>17</sub>N<sub>3</sub>O<sub>6</sub>,<sup>6)</sup> mp 177–178°C (from EtOH),  $[\alpha]_D^{20} -40.2^\circ$  (*c* 1, MeOH); 2-*C*-(Diethoxycarbonyl)methyl derivative **8**, C<sub>21</sub>H<sub>27</sub>NO<sub>10</sub>,<sup>6)</sup> mp 109–110°C (from EtOH),  $[\alpha]_D^{20} -61.2^\circ$  (*c* 1, MeOH).

Under similar conditions cyclohexanone was shown to be inactive as a nucleophile, but its morpholino enamine was made to react with **2** without a catalyzer to afford adduct **9** in 76% yield, C<sub>24</sub>H<sub>32</sub>N<sub>2</sub>O<sub>7</sub>,<sup>6)</sup> mp 145.5–146°C (from EtOH),  $[\alpha]_D^{20} -135^\circ$  (*c* 0.5, CHCl<sub>3</sub>).

Structures of the products obtained were deduced from NMR-data: The fact that  $J_{1,2}=0$  (dihedral angle of H<sup>1</sup>, H<sup>2</sup>=100°) and  $J_{4,5}=9.5$  Hz (diaxial of H<sup>4</sup>, H<sup>5</sup>) in **5** permits only a distorted  $\beta$ -*allo* structure, as shown in the figure. Large values of coupling constants ( $\geq 8$  Hz) of H<sup>1</sup>-H<sup>5</sup> in **6–9** indicate their  $\beta$ -*gluco* configurations.

TABLE 1. 100 MHz NMR SPECTRA IN CDCl<sub>3</sub> (TMS as an internal standard)

Compound	Chemical shifts in $\tau$					Coupling constants			
	H <sup>1</sup>	H <sup>2</sup>	H <sup>3</sup>	H <sup>4</sup>	$-\text{CH} \begin{smallmatrix} \text{O} \\ \diagup \diagdown \end{smallmatrix}$	$J_{1,2}$	$J_{2,3}$	$J_{3,4}$	$J_{4,5}$
<b>5</b>	5.21	6.25	—	4.98	4.30	0	—	—	9.5
<b>6</b>	5.30	?	5.00	5.87	4.47	8	10.5	9.5	9.5
<b>7</b>	5.41	7.16	5.12	5.79	4.44	8	10.5	9.5	9.5
<b>8</b>	4.87	7.10	4.97	?	4.46	8	11.5	10	?
<b>9</b>	5.51	?	5.01	5.87	4.48	8	10.5	10.5	9.5

1) T. Nakagawa, T. Sakakibara and S. Kumazawa, *Tetrahedron Lett.*, **1970**, 1645.

2) H. Newman and R. B. Angier, *Chem. Commun.*, **1969**, 369.

3) Value in water at 25°C [R. G. Pearson and R. L. Dillon, *J. Amer. Chem. Soc.*, **75**, 2439 (1953)].

4) In this case sodium hypochlorite is likely to be-

have as a basic catalyzer. The reaction proceeds also in the presence of a trace of sodium hydroxide instead of sodium hypochlorite, to give **6** in 86% yield.

5) H. H. Baer and T. Neilson, *Can. J. Chem.*, **43**, 840 (1965).

6) All the new compounds give satisfactory results in elementary analyses.